

Daily Tutorial Sheet [DTS] | Hydrocarbons

Level-1		DTS 1-5
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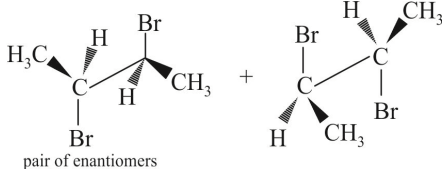
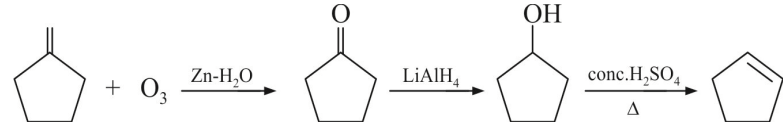
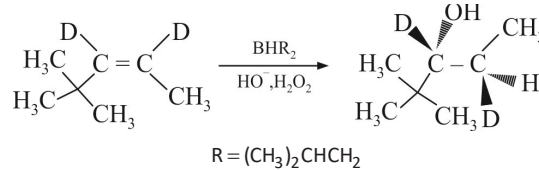
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
C	C	C	C	A	C	B	C	D	A	B	C	D	C	C
16	17	18	19	20	21	22	23	24	25	26	27	28	29	30
B	D	B	C	D	A	B	A	A	C	A	A	A	C	D
31	32	33	34	35	36	37	38	39	40	41	42	43	44	45
A	C	D	B	B	D	B	D	D	D	B	A	D	B	C
46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
D	A	AC	B	D	A	D	B	A	C	D	C	A	B	B
61	62	63	64	65	66	67	68	69	70	71	72	73	74	75
BCD	A	B	C	A	B	AB	B	A	ABCD	A	B	D	B	C

Level-2		DTS 6-11
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76	77	78	79	80	81	82	83	84	85	86	87	88	89	90
B	A	B	A	B	A	D	A	ABCD	A	B	A	B	D	D
91	92	93	94	95	96	97	98	99	100	101	102	103	104	105
C	C	C	A	C	C	D	B	C	C	BD	B	C	ABCD	A
106	107	108	109	110	111	112	113	114	115	116	117	118	119	120
D	D	D	D	C	C	D	B	B	A	C	B	A	B	C
121	122	123	124	125	126	127	128	129	130	131	132	133	134	135
B	A	ABD	ABC	B	2	2	6	7	5	3	7	2	3	4
136	137	138	139	140										
3	3	7	9	3										

JEE Main (Archive)		DTS 1-3
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1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
C	B	A	B	B	A	B	B	B	A	C	A	D	B	D
16	17	18	19	20	21	22	23	24	25	26	27	28	29	30
C	B	A	D	B	C	A	C	D	C	B	B	C	D	D
31	32	33	34	35	36	37	38	39	40	41	42	43	44	45
B	C	B	D	B	A	C	B	B	A	C	D	D	13.00	D
46														
B														

JEE Advanced (Archive)			DTS 1-8						
1.	Tollen's reagent	2.	B	3.	D	4.	Ethyne		
5.	$(\text{CH}_3)_2\text{C}=\text{CH}-\text{CH}_3$	6.	$\text{C}_2\text{H}_4 \xrightarrow[\text{CCl}_4]{\text{Br}_2} \xrightarrow[2\text{eq.}]{\text{NaNH}_2} \text{HC}\equiv\text{CH}$	7.	D				
8.	$3\text{C}_3\text{H}_6 + 2\text{KMnO}_4 + 4\text{H}_2\text{O} \longrightarrow 3\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{OH} + 2\text{MnO}_2 + 2\text{KOH}$	9.	False	10.	C				
11.	C	12.	B	13.	2-butyne				
14.	(i) Free radical mechanism	(ii)	Markovnikov's rule	15.	A				
16.	(i) HBr/peroxide	(ii)	Due to acidic nature of terminal alkyne.	17.	A				
18.	sp^3	19.	$\text{H}-\text{C}\equiv\text{C}-\text{H} + \text{Na} \xrightarrow{\text{heat}} \xrightarrow{\text{CHI}_3} \text{CH}_3-\text{C}\equiv\text{CH} \xrightarrow[\text{HgSO}_4]{\text{H}_2\text{SO}_4} \text{CH}_3-\overset{\text{O}}{\underset{\text{ }}{\text{C}}}-\text{CH}_3$ acetone						
20.	Tollen's reagent or fehling's test	21.	B						
22.	(A) $(\text{CH}_3)_2\text{CHCH}(\text{Cl})\text{CH}_2\text{CH}_3$	(B) $(\text{CH}_3)_2\text{CHCH}=\text{CHCH}_3$	(C) $(\text{CH}_3)_2\text{C}=\text{CHCH}_2\text{CH}_3$						
23.	AD	24.	B	25.	C	26.	D		
27.	$3\text{C}_2\text{H}_4 + 2\text{KMnO}_4 + 4\text{H}_2\text{O} \longrightarrow 3\text{HOCH}_2\text{CH}_2\text{OH} + 2\text{MnO}_2 + 2\text{KOH}$	28.	Baeyer's reagent.	29.	B				
30.	H_2SO_4 & HgSO_4	31.	Ethene						
32.	(i) (A) $\text{CH}_3\text{CH}=\text{CHCH}_3$	(ii) (B) $(\text{CH}_3)_2\text{C}=\text{CH}_2$	33.	(A) O_3 , (B) KO_3 , (C) O_2					
34.	 pair of enantiomers		35.	$(\text{CH}_3)_2\text{CH}-\text{CH}=\text{CH}_2$					
36.	(X) $(\text{CH}_3)_2\text{CHC}(\text{Cl})(\text{CH}_3)_2$	(Y) $(\text{CH}_3)_2\text{CHC}(\text{CH}_3)=\text{CH}_2$	(Z) $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$						
37.	(a) $\text{CH}_3\text{CH}(\text{Br})\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$	(b) $\text{CH}_3\text{CH}(\text{Br})\text{C}(\text{OH})(\text{CH}_3)\text{CH}_2\text{CH}_3$	(c) $\text{CH}_3(\text{C}_2\text{H}_5)_2\text{COH}$						
38.	A	39.	D	40.	1, 4-dibromo-2-butene	41.	3-Hexene		
42.	C	43.	C	44.	A	45.	(A) $\text{C}_6\text{H}_5\text{CH}(\text{Br})\text{CH}_2\text{Br}$ (B) $\text{C}_6\text{H}_5\text{C}\equiv\text{CCH}_3$		
46.	A	47.	D						
48.	(a)  (b)  $\text{R} = (\text{CH}_3)_2\text{CHCH}_2$								
49.	D	50.	B	51.	C	52.	C	53.	B
54.	B	55.	B	56.	A	57.	Syn-addition across $\text{C}\equiv\text{C}$		
58.	B	59.	D	60.	X : 3-Hexyne ; Y : cis-3-Hexene ; Z : Meso-hexane-3, 4-diol				
61.	B	62.	C	63.	B	64.	A	65.	B
66.	A	67.	A	68.	B	69.	C	70.	D

71.	A	72.	$\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH} = \text{CH}_2$	73.	B	74.	B
75.	B	76.	C	77.	B	78.	B
80.	D	81.	A	82.	B	83.	A
85.	D	86.	C	87.	D	88.	B
90.	D	91.	5	92.	C	93.	8
95.	B	96.	A	97.	B	98.	D
100.	BD	101.	D	102.	AB	103.	B
105.	BCD	106.	A	107.	B	108.	D
110.	CD	111.	D	112.	A	113.	B
115.	CD	116.	135			114.	A